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The unit-cell constants of some PuNi ${ }_{3}$-type compounds*. By A. E. Dwight, Metallurgy Division, Argonne National Laboratory, Argonne, Illinois 60439, U.S.A.
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Crystal structure data, including unit-cell constants and observed intensities obtained from powder patterns, are presented for $\mathrm{UCo}_{3}$ and other $\mathrm{PuNi}_{3}$-type compounds.

Earlier work (Elliott, 1965) on the U-Co system indicated that the compound $\mathrm{UCo}_{3}$ was formed at $\sim 880^{\circ} \mathrm{C}$ by a peritectoid reaction. The crystal structure of the compound was not identified. Our investigation shows that $\mathrm{UCo}_{3}$ is isostructural with $\mathrm{PuNi}_{3}$, which was reported (Cromer \& Olsen, 1959) to be rhombohedral, $R \overline{3} m$.

The $\mathrm{UCo}_{3}$ compound investigated was prepared by arc melting electrolytic uranium and $99.99 \%$ Co under an argon-helium atmosphere. The button was homogenized at $800^{\circ} \mathrm{C}$ for four days, and a powdered specimen was annealed at $800^{\circ} \mathrm{C}$ for 18 hours. X-ray diffraction patterns were obtained with a Debye-Scherrer camera and chromium $K \propto$ radiation. Although the structure is rhombohedral, the diffraction pattern was indexed on the basis of the related hexagonal unit cell. The unit-cell constants were cal-

[^0]Table 1. $\mathrm{UCo}_{3}$ structural data based on a hexagonal unit cell Cr radiation ( $K \alpha_{1}=2 \cdot 28962 \AA$ )

| $h k l$ | $d_{\mathrm{o}}$ | $d_{\mathrm{c}}$ | $I_{\mathrm{o}}$ |
| :--- | :--- | :--- | :--- |
| 1001 | $4.105 \AA$ | $4 \cdot 138 \AA$ | $w$ |
| 006 |  | 4.052 |  |
| 102 |  | 3.969 |  |
| 104 |  | 3.455 |  |
| 105 |  | 3.179 |  |
| 009 |  | 2.702 |  |
| 107 | 2.66 | 2.677 | $m$ |
| 108 | 2.45 | 2.462 | $m$ |
| 110 | 2.412 | 2.425 | $m$ |
| 113 |  | 2.329 |  |
| $1,0,10$ |  | 2.104 |  |
| 201 | 2.09 | 2.092 | $m$ |
| 116 | 2.075 | 2.081 | $s$ |
| 202 |  | 2.069 |  |
| 00,12 | 2.022 | 2.026 | $v w$ |
| 204 | 1.98 | 1.985 | $v w$ |
| $1,0,11$ | 1.95 | 1.956 | $v w$ |
| 205 | 1.925 | 1.928 | $w$ |

Table 1 (cont.)

| hkl | $d_{0}$ | $d_{\text {c }}$ | $I_{0}$ |
| :---: | :---: | :---: | :---: |
| 119 | 1.80 | 1.805 | $w$ |
| 207 |  | 1.797 |  |
| 208 |  | 1.728 |  |
| 1,0,13 |  | 1.709 |  |
| 00,15 | 1.619 | 1.621 | $w$ |
| 1,0,14 | 1.604 | 1.605 | $w$ |
| 20,10 |  | 1.589 |  |
| 211 | 1.581 | 1.584 | $w$ |
| 212 |  | 1.574 |  |
| 1,1,12 |  | 1.555 |  |
| 214 |  | 1.536 |  |
| 2,0,11 |  | 1.522 |  |
| 215 |  | 1.509 |  |
| 217 | 1.443 | 1.444 | $m$ |
| 1,0,16 | 1.429 | 1.429 | vw |
| 218 | 1.407 | $1 \cdot 407$ | $w$ |
| 300 | $1 \cdot 400$ | 1.400 | $m$ |
| 2,0,13 | $1 \cdot 397$ | $1 \cdot 397$ | $w$ |
| 303 |  | 1.379 |  |
| 1,0,17 |  | $1 \cdot 354$ |  |
| 00,18 |  | $1 \cdot 351$ |  |
| 1,1,15 | $1 \cdot 347$ | $1 \cdot 348$ | $m$ |
| 20,14 | $1 \cdot 338$ | $1 \cdot 338$ | $m$ |
| 2,1,10 |  | 1.329 |  |
| 306 | $1 \cdot 322$ | 1.323 | $s$ |
| 2,1,11 | $1 \cdot 289$ | 1.289 | m |
| 309 | $1 \cdot 2425$ | 1-2429 | $w$ |
| 2,0,16 |  | $1 \cdot 2312$ |  |
| 1,0,19 | 1.225 | $1 \cdot 2242$ | $w$ |
| 220 | 1.2123 | $1 \cdot 2123$ | $s$ |
| 2,1,13 |  | $1 \cdot 2103$ |  |
| 223 |  | $1 \cdot 1990$ |  |
| 2,0,17 | $1 \cdot 182$ | $1 \cdot 1822$ | $m$ |
| 1,1,18 |  | $1 \cdot 1801$ |  |
| 2,1,14 | 1-1718 | $1 \cdot 1717$ | $s$ |
| 1,0,20 |  | 1.1679 |  |
| 311 | $1 \cdot 1632$ | $1 \cdot 1634$ | $m$ |
| 226 |  | $1 \cdot 1615$ |  |
| 312 |  | $1 \cdot 1594$ |  |
| 0,0,21 | $1 \cdot 158$ | $1 \cdot 1579$ | $m$ |
| 3,0,12 |  | $1 \cdot 1518$ |  |

Table 2. Unit-cell constants of PuNi ${ }_{3}$-type compounds based on the related hexagonal unit cell

| Compound | $a$ | $c$ | $c / a$ | Vol $/ M$ | Heat <br> treatment* |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{UCo}_{3}$ | $4.8492(2)$ | $24.317(1)$ | 5.02 | 55.03 |  |
| $\mathrm{GdFe}_{3}$ | $5.1654(7)$ | $24.707(2)$ | 4.78 | 63.43 | $(a)$ |
| $\mathrm{HoFe}_{3}$ | $5.1097(3)$ | $24.526(1)$ | 4.80 | 61.6 | $(b)$ |
| $\mathrm{TbCo}_{3}$ | $5.0156(3)$ | $24.424(1)$ | 4.87 | 59.13 | $(c)$ |
| $\mathrm{LuCo}_{3}$ | $4.955(2)$ | $24.101(9)$ | 4.86 | 56.94 | $(d)$ |
| $\mathrm{HoNi}_{3}$ | $4.958(3)$ | $24.33(1)$ | 4.91 | 57.4 | $(a)$ |
| $\mathrm{ErNi}_{3}$ | $4.948(2)$ | $24.27(1)$ | 4.91 | 57.18 | $(a)$ |
| $\mathrm{TmNi}_{3}$ | $4.937(2)$ | $24.213(9)$ | 4.90 | 56.79 | $(a)$ |

* (a) Button homogenized 3 days at $1000^{\circ} \mathrm{C}$; powder not annealed.
(b) Button homogenized 19 hr at $900^{\circ} \mathrm{C}$; powder not annealed.
(c) Button homogenized 5 days at $500^{\circ} \mathrm{C}$; powder annealed 3 hr at $500^{\circ} \mathrm{C}$
(d) Button homogenized 5 days at $900^{\circ} \mathrm{C}$; powder annealed $21 \mathrm{hr} 900^{\circ} \mathrm{C}$.
culated by the computer program of Mueller, Heaton \& Miller (1960), and the $d$ spacings were obtained by the program of Mueller, Meyer \& Simonsen (1962). The density, measured by the immersion method, is $12.44 \mathrm{~g}_{\mathrm{cm}} \mathrm{cm}^{-3}$, and the X-ray density is $12.52 \mathrm{~g} . \mathrm{cm}^{-3}$. The observed and calculated $d$ spacings and observed intensities are listed in Table 1. The observed intensities are in satisfactory agreement with the observed and calculated $F^{2}$ values reported by Bertaut, Lemaire \& Schweizer (1965) for $\mathrm{HoCo}_{3}$. The unit-cell constants of $\mathrm{UCO}_{3}$ and of several isostructural rare-earth compounds are listed in Table 2. In this Table the Figure in parentheses is the least-squares standard error of the last significant digit. The rare-earth compounds received various heat treatments (see Table 2), but all X-ray patterns were equally well resolved.

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Crystallographic data on disubstituted symmetric ureas. By S. V.Deshapande, Physics Department, Sardar Patel University, Vallabh Vidyanagar, Gujarat State, India and C.C.Meredith and R.A.Pasternak,* Stanford Research Institute, Menlo Park, California 94025, U.S.A.
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The unit-cell dimensions and space groups of six disubstituted, symmetric ureas ( RHN$)_{2} \mathrm{C}=\mathrm{O}$ have been established. The substituents R in this series were phenyl, $p$ - and $m$-tolyl, $m$ - and $o$-chlorophenyl and $p$-anisyl. Similar packing of the molecules in the unit cells is suggested by the data.

We report here the unit cells and space groups of six disubstituted symmetric ureas, $(\mathrm{RHN})_{2} \mathrm{C}=\mathrm{O}$, with $\mathrm{R}=$ phenyl, $p$ - and $m$-tolyl, $m$ and $o$-chlorophenyl and $p$-anisyl.

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Needle crystals were obtained for all the compounds by slow evaporation of their solutions in $96 \%$ ethanol. They all showed good cleavage along two directions parallel to the needle axis and no cleavage perpendicular to it. Preliminary unit-cell dimensions were derived by indexing rotation photographs around the needle axis which was

Table 1. Crystal data for urea derivatives

\begin{tabular}{|c|c|c|c|c|c|c|c|c|c|}
\hline \& Molecular weight \& $$
a^{\mathrm{A}}
$$ \& $$
\begin{aligned}
& \text { xial lengtl } \\
& b
\end{aligned}
$$ \& $c$ \& Measured density \& $$
\begin{gathered}
\text { Number } \\
\text { of } \\
\text { molecules }
\end{gathered}
$$ \& Space group \& Crystal system \& Remarks <br>
\hline Urea* \& 60.06 \& 5.66 A \& 5.66 Å \& 4.72 Å \& $1.33 \mathrm{~g} . \mathrm{cm}^{-3}$ \& 2.02 \& $P^{4} 2_{1}{ }^{\text {m }}$ \& Tetragonal \& <br>
\hline Diphenylurea \& $212 \cdot 25$ \& 10.51 \& 11.73 \& 9.07 \& 1.23 \& 3.90 \& $P 2_{1}$ cn \& Orthorhombic \& <br>
\hline Di-p-tolylurea \& $240 \cdot 29$ \& 9.85 \& 27.77 \& $4 \cdot 66$ \& $1 \cdot 26$ \& 4.02 \& $P{ }_{2}{ }_{1} a$ \& Orthorhombic \& <br>
\hline Di-m-tolylurea \& $240 \cdot 29$ \& 9.72 \& 14.56 \& $4 \cdot 60$ \& 1.25 \& 2.04 \& $$
\begin{aligned}
& P 2_{1} 2_{1} 2 \\
& \left(P 2_{1} 2_{1} 2_{1}\right)
\end{aligned}
$$ \& Orthorhombic \& <br>
\hline Di- $m$-chlorophenylurea \& 281-14 \& 9.72 \& 14.36 \& 4.55 \& 1.47 \& 2.00 \& $$
\begin{aligned}
& P 2_{1} 2_{1} 2 \\
& \left(P 2_{1} 2_{1} 2_{1}\right)
\end{aligned}
$$ \& Orthorhombic \& <br>
\hline Di-o-chlorophenylurea \& 281.14

272.29 \& $\begin{aligned} & 23.00 \\ & \\ & \\ & \\ & \gamma^{*}=\end{aligned}$ \& $23 \cdot 20$
$81^{\circ} \mathrm{O}$ \& $4 \cdot 64$ \& 1.48
1.35 \& 7.85
7.95 \& Pban

P1 or PT \& Orthorhombic \& Poor crystals. $h=2 n$, very weak for $h 00, h 01$. $k=2 n$, very weak for $0 k 0,0 k 1$ <br>

\hline Di-p-anisylurea \& $272 \cdot 29$ \& \[
$$
\begin{aligned}
& \left(\gamma^{*}=\right. \\
& 21 \cdot 20 \\
& \left(=d_{100}\right)
\end{aligned}
$$

\] \& \[

$$
\begin{aligned}
& \left.81^{\circ}\right) \\
& 13 \cdot 38 \\
& \left(=d_{010}\right)
\end{aligned}
$$
\] \& $9 \cdot 31$ \& 1.35 \& $7 \cdot 95$ \& $P 1$ or PT \& ${ }^{\text {Triclinic }}$ \& Odd layer lines on rotation about $c$ axis very weak. $h k 0$ with $h+k$ odd absent; $h$ and $k$ odd very weak. <br>

\hline
\end{tabular}


[^0]:    * This work performed under the auspices of the U.S. Atomic Energy Commission.

